

Research Article

Study of the impact of heavy metals and physico-chemical parameters on the quality of the wells water Around Yedgaon Dam area Junner, Dist: Pune.

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ABSTRACT

To analyze the impact of heavy metals on well and soil water quality from the Yedgaon Dam Region, we studied the physicochemical parameters (pH, electrical conductivity, Organic matter, chlorides, sodium, potassium, nitrates, ammonium, sulphates) as well as the evaluation of the degree of metal contamination (iron, zinc, chromium, copper and lead) present in the samples. The analyses carried out revealed low levels of organic matter (MO: max = 1.5 mg /l and mineral matter (conductivity: max = 775 ms / cm).) Analysis of waters of wells and soil showed that grades In NO₃⁻ present an average concentration of less than 40 mg /l with temporal variations of remarkable amplitude (in accordance with the standards suggested by the World Health Organization and the Moroccan standards). The average chloride contents do not exceed 785 mg /l (suggested by Moroccan standard). As for nitrous nitrogen, the recorded values showed very low contents and were found lower than the Moroccan standards (0.1 mg/l). Concerning the heavy metals, the examined results showed a significant increase in the concentration of lead in the seven wells, in particular well No. 1 (10.0 µg / l), and an elevation in iron and zinc concentration in the seven wells. The copper and chromium contents were relatively low. In general, the concentrations of these metals at the level of the different samples were found lower. In the case of soil analyses, the results showed a high concentration of iron in the seven soil locations, the contents of zinc, chromium and copper remained low, though the lead contents were considerably low.

KEYWORDS

Heavy Metals, Physico-chemical, Yedgaon Dam, Well water, Soil

1. INTRODUCTION

Heavy metals are the elements with a density greater than 5 g / cm³. They can be found in air, water and soil. These metals often interfere with the normal course of metabolic processes even in trace amounts [1], causing several diseases and act by accumulation effects [2]. Such metals often have a toxic effect, so their presence in the aquatic ecosystem poses risks to human health and causes harmful effects to living organisms [3]. The industry development has resulted in a significant increase of residues in natural waters and soils, especially those containing heavy metals. Therefore, the determination of the contents of metallic trace elements in water and soil is essential. In addition, legislation on environmental protection is becoming more and more stringent and encourages manufacturers to reduce the concentration of certain metal ions [4]. In this context, this paper deals with the impact of metals on the quality of well water and soil of the Yedgaon Dam region, which represent a source of production of these metals during various man made activities and Vignahar Sugar Factory located nearby study area.

2. MATERIALS AND METHODS

2.1. Description of the study area and Geographic location

Yedgaon Dam, is situated between Latitude 19°10' N and E Longitude 75°01' in Junner Tahasil, Pune District. Catchments Area of this Dam is 461 Sq. Km. At main river valley Krushna and sub valley Bhima. The rainfall in this area is 25 to 37 inch per year. The water storage capacity of this Dam is 3300 Quebec Feet. Yedgaon Dam, is an earthfill and gravity dam on Kukadi river near Junnar, Pune district in the state of Maharashtra in India. The height of the dam above its lowest foundation is 29.74 m (97.6 ft) while the length is 4,511 m (14,800 ft). The volume content is 1,004 km³ (241 cu mi) and gross storage capacity is 93,430.00 km³ (22,415.05 cu mi). The dam is located in the Ghod basin and is part of the Kukadi project, which constructed five dams in the region. Other dams included in this project are Wadaj Dam, Manikdoh Dam, Dimbhe Dam and Pimpalgaon Joge Dam. It is situated near to Pune-Nasik National Highway 50. It is 77 km north of Pune, 132 km south of Nasik and around 176 km east of the state capital, Mumbai [5] Sector and Geological map and percentage of water used around study area are shown in Figure 1, 2 and 3.



Figure 1. Sector of study around Yedgaon Dam.

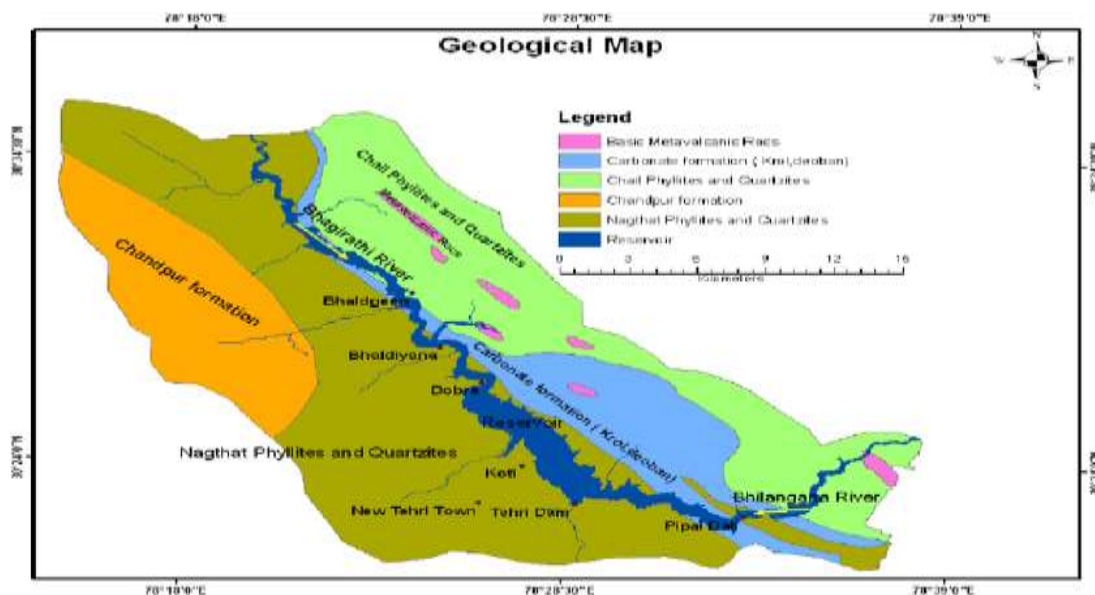


Figure 2. Synthetic geological map of the study area

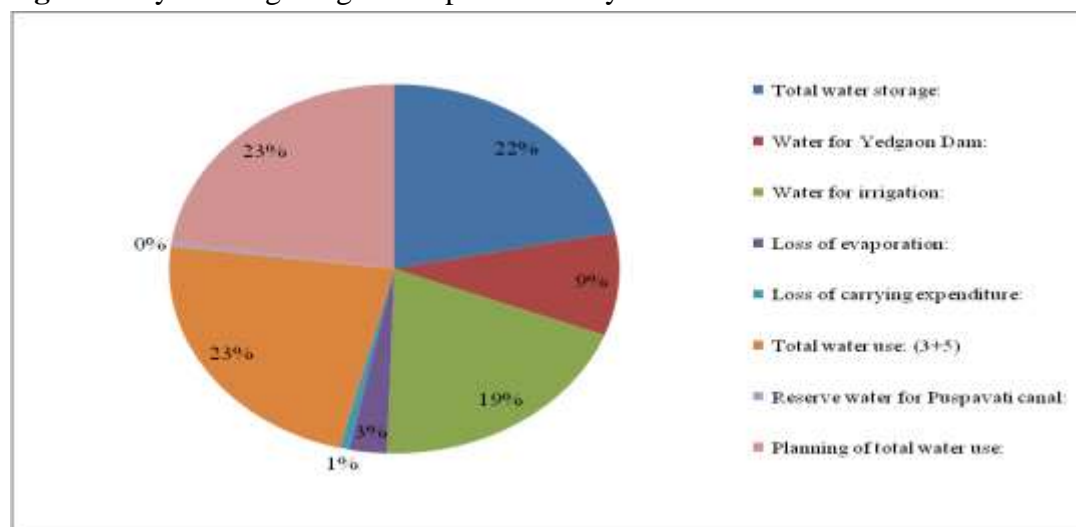


Figure 3. Water Use of Yedgaon Dam

2.1.2. Climate framework

The study shows the maximum temperature of an average below 40°C day for every month for Junnar. Hot days and cold nights show the average of the hottest day and coldest night of each month of the last 30 years. Wind speeds are not displayed per default.

2.1.3. Choice and preparation of samples

To estimate the degree of metallic contamination of waters of wells and ground which surround the Dam, 7 wells and 7 locations from various ground soil locations near each well were exploited. The Location of sampling points around the Yedgaon Dam is reported in Figure 4 and Table 1.

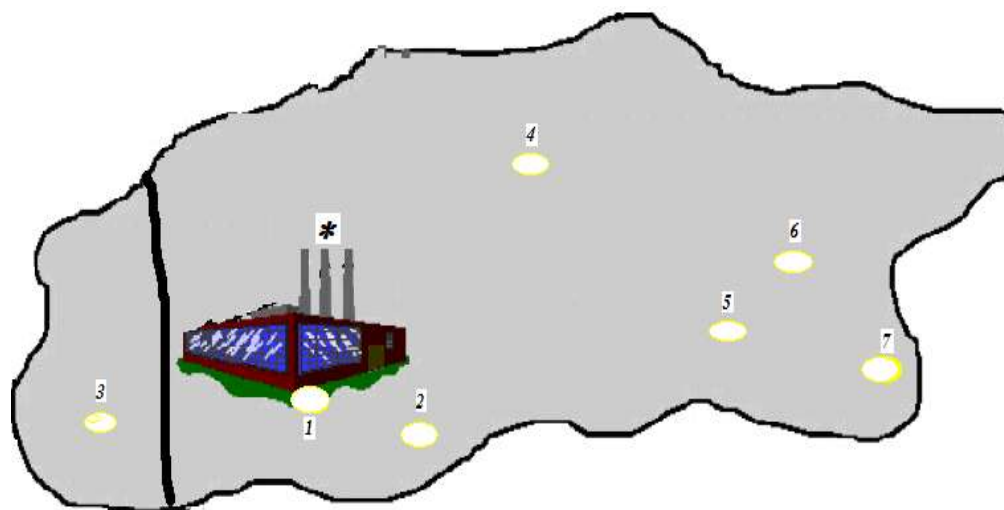


Figure 4. Location of sampling points: **—** : Road to Shivneri Fort and * : Vignahar Sugar Factory.

Table 1. Location of sampling points around the Yedgaon Dam.

Wells	Position by supplying the plant	Distance
1	Factory	-
2	South West	150 m
3	South East	300 m
4	North West	1500 m
5	West	3000 m
6	West	3500 m
7	West	4000 m

All wells have a lid and are on farms except well No. 1 which is located within the study area. For physicochemical analyses: water samples are taken from plastic bottles, well rinsed with distilled water and with the water to be analyzed, at the time of collection. For heavy metal analyses: water samples are taken from 50 ml polyethylene conical tubes; the samples are acidified with 2% nitric acid (HNO₃). Soils were removed from the surface (5 cm thick) using a non-metallic tool and stored in clean plastic bags and kept at low temperature. The soils were dried at 105 °C. In the oven for 24 hours, then ground and homogenized. The samples are stored in special plastic bags to avoid any risk of contamination. Labelling is rigorous to avoid any confusion about the identity of the samples [6].

2.1.4. Analytical methods

2.1.4.1. Principle and procedure of physicochemical parameters

The analyses of the various physical-chemical parameters were carried out according to the standard method [7]. Temperature, hydrogen potential (pH) and electrical conductivity were measured at the time of sample collection. PH was measured with the Portable pH-meter field, the type (WTW pH 330i / SET), and conductivity with conductivity meter, type (WTW Cond

330i / SET). The temperature was measured by an integrated thermometer in the measurement of conductivity and pH-meter. Nitrate (NO₃⁻), Nitrite (NO₂⁻), Ammonia nitrogen (NH₄⁺) and sulphate (SO₄²⁻) are determined by a colorimetric method using a UV / visible spectrophotometer. Calcium (Ca₂⁺) and magnesium (Mg₂⁺) were determined by the volumetric method with EDTA (ethylenediamine acid), the oxidizability (organic matter: MO) was determined by high oxidation Temperature in acid medium, Total alkalinity (AT) and bicarbonate (HCO₃⁻) are analyzed by volumetric determination with 0.1N HCl. Sodium (Na⁺) and potassium (K⁺) are determined by exciting atoms by flame type photometer (AFP-100).

2.1.4.2. Principle and procedure of metallic traces in Waters and soil

The analysis of metals in water and soil is carried out by atomic absorption spectrophotometer. Metal contents such as Pb, Cu and Cr are determined directly by Atomic Absorption Spectrophotometer with graphite furnace (AAS-GF) type Varian 240, Zeeman. Iron and zinc were measured by the AAS flame, the Varian Spectrophotometer type with atomic absorption four (SAAF) type Varian, 240 Sequential fast. The determination of total metal trace elements in the soil samples is carried out by mineralization in closed Teflon "bombs" using a combination of hydrofluoric acid (HF) and regal water (HNO₃: HCl, 1: 3 V / V) to decompose the sample (IAEA method).

3. RESULTS AND DISCUSSIONS

The result of Analysis of physicochemical parameters is reported in Table 2.

Table 2. summarizes all the results of the physical-chemical parameters found.

Site	P H	E.C. (µs/cm)	OM (mg/l)	AN (mg/l)	TA (F°)	TA C (F°)	TH (F°)	Ca ⁺⁺ (mg/l)	Mg ⁺⁺ (mg/l)	Na ⁺ (mg/l)	K ⁺ (mg/l)	HCO 3 ⁻ (mg/l)	Cl ⁻ (mg/l)	NO3 ⁻ (mg/l)	SO ₄ ²⁻ (mg/l)
Wel 11	7.8	770	1.4	0.00	0	30.	25.8	20.0	50.0	87.8	2	378.	46.1	33.9	36.1
Wel 12	7.6	750	0.77	0.00	0	27.	350	22.40	71.9	33.3	1	341.	74.5	32.5	32.8
Wel 13	7.7	674	0.77	0.00	0	30.	35.2	16.80	76.8	21.4	1	420.	46.1	35.8	27.9
Wel 14	7.4	768	0.510.	0.00	0	29.	37.1	41.65	65.1	20.7	1	360.	53.2	41.3	38.6
Wel 15	7.5	710	0.58	0.00	0	29.	33.7	15.20	72.9	22.9	1	360.	49.5	36.0	29.6
Wel 16	7.4	722	0.50	0.00	0	28.	33.5	28.0	64.6	25.1	1	353.	49.5	29.9	39.5
Wel 17	7.5	702	0.49	0.00	0	29.	34.0	32.85	62.7	25.1	1	359.	53.2	26.1	34.5

3.1.1. Hydrogen potential

The pH of the water measures the concentration of the H^+ protons contained in the water. It summarizes the stability of the balance between the different forms carbonic acid and is linked to the buffer system developed by carbonates and bicarbonates [7]. The pH values of the water samples vary between 7.30 and 7.85, which qualifies them nearly neutral. The 100% of the water analyzed conforms to the quality of the water used for human consumption.

3.1.2. Conductivity

Electrical conductivity refers to the ability of the water to conduct an electrical current. It is determined by the dissolved contents, the ionic charge, the ionization capacity, the mobility and the temperature of the water. Thus, the measurement of the conductivity makes it feasible to appreciate the quantity of the dissolved salts in the water and consequently its mineralization [8]. All the values of the conductivities found (between 675 and 775 $\mu s / cm$) are lower than the maximum permissible value indicates that water is poorly mineralized.

3.1.3. Organic matter

Organic matter is a parameter that informs the state of water in the well and, on the other hand, promotes the growth of microorganisms that degrade organic matter. In general, low values favour the development of pathogenic germs [9]. For all the samples, the organic matter contents vary from 0.50 to 1.5 mg/l during the study period and the results obtained show that all the recorded values are below the maximum permissible value (AMV=2mg/l). The excess of the organic matter recorded in Well No.1 may be related either to domestic releases or to the industrial activities.

3.1.4. Ammonia nitrogen

Ammoniacal nitrogen is one of the links in the complex nitrogen cycle in its primitive state. Ammonia is a water soluble gas. Ammoniacal nitrogen is therefore a good indicator of pollution of watercourses by organic discharges of agricultural, domestic or industrial origin [10]. In our study area, no trace of this element was reported.

3.1.5. Nitrous nitrogen

A concentration of nitrous nitrogen greater than 0.10 mg /l can lead to the suspicion of a supply of water rich in decomposing organic matter [11]. This content should not be exceeded in the case of water of deep origin. In our study, nitrites were detected in the waters of the 7 wells, have very low levels and are below normal (0.1 mg / l)

3.1.6. Title of Alkalinity (TA) and Title of Complete Alkalinity (TAC)

From the results obtained, we note that the TA = 0 in the seven wells analyzed, that is to say that, TAC = $[HCO_3^-]$. The results found show that the major alkaline element, which exists, is the bicarbonate HCO_3^- , varies from 341.5 to 420.5 mg / l. These results are high because of the calcareous nature that characterizes the soil of the study region [12].

3.1.7. The total hardness

In most cases the hardness is mainly due to the Ca^{2+} and Mg^{2+} ions, to which Fe^{2+} , Mn^{2+} and Sr^{2+} ions are sometimes added. The variation in total hardness (TH) observed in the waters of the different wells can be related to the nature of the soil in the area. The values recorded in our study exceed the maximum permissible value (AMV = 30 ° F) according to Moroccan standards

(NM 03.7.001 March 2006), except Well No. 1. Hence, the nature of groundwater in the region are hard due to the calcareous nature of the parent rock which is rich in Ca_2^+ and Mg_2^+ [12].

3.1.8. Chloride

Chlorides are widely distributed in nature, usually in the form of salts of sodium (NaCl) and potassium (KCl). They are often used as pollution indicators [13]. In addition, they exist in all waters in variable concentrations. The chloride contents of the analyzed water samples (Table 2) show values ranging from 46.10 mg/l to 74.55 mg/l. The results show that the chloride once nitration is less than 750 mg / l .

3.1.9. Nitrates

Nitrates are found in excess, they are an incontestable marker of pollution. At the level of the studied waters, the nitrate contents vary between 26.15 and 41.35 mg / l, according to the standards, VMA = 50 mg . This alteration of the quality of sinks by nitrates could be attributed to the natural degradation of organic matter or to nitrogenous fertilizers [13].

3.1.10. Sulphates

Sulphates result from the streaming or infiltration into gypsum grounds. They also result from the activity of some bacteria (chlorothio-bacteria, rhodothi-obacteria, etc.). This activity can oxidize hydrogen sulphide (H_2S), toxic sulphate [14], etc. According to the results of the samples analyzed (FIG. 6), the values recorded from 27.95 to 39.50 mg / l remain below the maximum permissible value (VMA = 400 mg / l) according to the Moroccan standard relating to the quality of water intended for Production of drinking water. All results are reported in Table 2.

3.2. Analysis of metallic contamination of water and soil by heavy metals

The results obtained by the analysis of the total zinc, iron, chromium, copper and lead levels in the water and soil samples are shown in Tables 3 and 4. The results of the analyses of trace metallic elements in the examined water samples show an important increase in the concentration of lead in the seven wells, in particular well No. 1 ($9.95 \mu\text{g} / \text{l}$), and a rise in concentration of iron and zinc in the seven wells, while the copper and chromium contents are relatively low. Generally, concentrations of these metals in the various samples are below the Moroccan standards. In the case of soil analyses, the results show a high concentration of iron in the seven soil locations, the contents of zinc, chromium and copper remain low, but the contents of lead are considerably low [15]. The results are summarized in Table 3 and 4.

Table 3. Change in heavy metals in water samples.

Nature of sample	No. of well	Metal ($\mu\text{g} / \text{l}$)				
		Zn	Fe	Cr	Cu	Pb
Water	1	275	9	2.74	24.84	9.95
	2	586.7	37	2.5	3.34	7.48
	3	61.2	14	3.07	21.3	8.73
	4	17.1	29	3.02	0.1	7.16
	5	16.2	154	2.96	0.32	5.15
	6	22.9	63	2.73	1.58	6.69
	7	32.8	57	2.43	1.42	6.74

Table 4. Change in heavy metals in soil samples.

Nature of sample	No. of well	Metal ($\mu\text{g/l}$)				
		Zn	Fe	Cr	Cu	Pb
Soil	1	57.70	22.10	80.50	48.35	2.80
	2	57.30	10.85	68.10	46.50	0.555
	3	46.0	12.30	38.50	38.50	0.560
	4	72.8	14.45	29.80	36.68	5.40
	5	46.85	10.75	10.0	29.50	0.70
	6	51.20	11.47	35.37	35.60	0.60
	7	64.615	23.21	25.50	68.20	1.25

3.2.1. Total zinc levels in the analyzed waters

The obtained results show that the total zinc contents in the analyzed waters are high, varying between 17.10 $\mu\text{g/l}$ and 586.107 $\mu\text{g/l}$, the values recorded in the different wells are lower than the maximum admissible value (AMV= 3 mg/l) according. It is also worth noticing that the highest contents are in the surroundings of the factory, which could be explained on one hand by the dust and the waste emitted by the factory.

3.2.2. Total zinc contents in the analyzed soils

The results obtained show that the total zinc contents in the analyzed soils are high, ranging from 46.00 $\mu\text{g/g}$ to 72.10 $\mu\text{g/g}$. The high soil levels analyzed could be explained by dust, waste and the sprays emitted by the factory, transported by the wind and the washing of the road [16].

3.2.3. Total iron contents in the analyzed waters

The obtained results show that the total iron contents in the analyzed waters is variable, varying between 8 and 155 $\mu\text{g/l}$. The values recorded in the different wells are lower than the maximum admissible value (AMV = 0.3mg / L). The variation of the iron contents at the wells analyzed could be explained by piezometry and the mineralogical characteristics of rocks [17].

3.2.4. Total iron contents in the analyzed grounds

The results obtained show that the total iron content in the soils analyzed is very high, especially in samples No. 1 and No. 7, ranging from 10.5 to 23.5 $\mu\text{g/g}$. The high soil levels analyzed could be explained by the dust, waste and aerosols emitted by the plant, the mineralogical characteristics of the soil, and the use of organic fertilizers for the enrichment of iron soils. Iron concentrations are higher at points 1 and 7. These results suggest that the physicochemical conditions of the soil (Redox potential, humidity, etc.) and morphological conditions (Presence of obstacles, flows, etc.); have been more favourable to the fixation of iron by the different components of the soil [18].

3.2.5. Total Chromium levels in the analyzed waters

The obtained results show that the total chromium contents in the analyzed waters are low, varying between 2.5 and 3.1 $\mu\text{g/l}$, the values recorded in the various wells are below the maximum admissible value (VMA = 50 $\mu\text{g/L}$). The low contents of the analyzed waters could be explained by the impermeability of the soil layers [19].

3.2.6. Total chromium contents in the analyzed soils

The results obtained show that the total chromium contents in the analyzed waters are high, ranging from 9.10 to 80.50 $\mu\text{g} / \text{g}$, and concentrations decrease with the increase in distance from the plant. This increase in chromium levels in the soils analyzed could be explained by the dust and aerosols emitted by the plant and transported by the wind that does not manage to throw them further, the mineralogical characteristics of the soil could be an important cause [19].

3.2.7. Copper contents in the analyzed waters

The results obtained show that the total copper contents in the analyzed waters are low, varying between 0.32 and 24.84 $\mu\text{g} / \text{l}$, the values recorded in the different wells are lower than the maximum admissible value (VMA = 2mg / L) according to Moroccan standards (NM 03.7.001 March 2006). The contents in well No. 1 and No. 3 are higher relative to other wells [20].

3.2.8. Total copper contents in the analyzed soils

The results obtained show that the total copper contents in the analyzed soils are high, ranging from 29.5 to 68.2 $\mu\text{g}/\text{g}$. This increase in Copper levels in the soils analyzed could be explained by the dust and aerosols emitted by the plant transported by the wind and by the use of organic fertilizers which enrich the soil copper [20].

3.2.9. Total lead levels in the analyzed waters

The results obtained show that the total levels of lead in the analyzed waters are very high, varying between 5.15 and 9.95 $\mu\text{g}/\text{l}$. The values recorded in the different wells are lower than the maximum acceptable value (VMA=10 $\mu\text{g}/\text{l}$). Such concentration may be due either to the leaching of the lead-rich soil or to industrial discharges from the Holcime plant. Their contents in the well No. 1 are too high and the maximum permissible value (AMV = 10 $\mu\text{g} / \text{l}$) has been reached [20].

3.2.10. Total lead levels in the analyzed soils

The results obtained show that the total levels of Lead in the analyzed soils are variable, ranging from 0.560 $\mu\text{g} / \text{g}$ to 5.5 $\mu\text{g} / \text{g}$. The levels of lead in the soil analyzed could be explained on the one hand by the dust and aerosols emitted by the plant transported by the wind and on the other hand, by the use of organic fertilizers which enrich the soils, which often contain lead contamination [20].

4. CONCLUSION

Through various kinds of emissions, the cement industry (similar all other industries), contributes to the alteration of air and soil quality, which would have repercussions on the various components of the receiving environment. The soil, which occupies a privileged place through its role as an interface between the atmosphere, the biosphere and the geosphere, is particularly sensitive to this pollution, along with the underground water. In the present work, the obtained results by the evaluation of the quantity of heavy studied metals (iron, zinc, chromium, copper and lead) accumulated in waters and grounds of the region of the factory Holcime, allowed us to discuss the following observations: An important presence in the soil for all the trace metal elements studied which presents a contamination of this soil that will have an impact on human health through agric-food products. A variable presence of the trace metallic elements studied in the waters does not exceed the maximum permissible values according to

Moroccan standards for the quality of the water for human consumption, but these levels may increase due to the activities of the plant, which have become increasingly important due to the increased demand for cement. At the end of this work, we propose the following recommendations to limit the pollution generated by the cement industry:

- Repair dust removal systems after each failure.
- Protection of pollution sources (chimneys, careers) by trees in order to reduce the effect of wind on dust.
- Use closed and well-sealed storage for fuel deposits and cement manufacturing materials to curb dust dispersion.
- Installation of new technologies such as electrostatic filter and bag filter on chimneys

5. REFERENCES

1. Mkadmi Y., Benabbi O., Fekhaoui M., Benakkam R., Bjjjou W., Elazzouzi M., Kadourri M., Chetouani A., (2018), Study of the Impact of Heavy Metals and Physico-chemical Parameters on the Quality of the Wells and Waters of the Holcim Area (Oriental region of Morocco), *Mater. Environ. Sci.*, 9(2), 672-679.
2. Jianrong Chen a, Shanmei Xiao, Xiaohua Wu, Keming Fang, Wenhan Liu., (2005), Determination of Lead in Water Samples by Graphite Furnace Atomic Absorption Spectrometry After Cloud Point Extraction, *J. Chen et al. / Talanta.*, 67,992.
3. Van de voorde, pinoy iv, l., courtijn e, and Francis Verpoort. (2005), Influence of Acetate Ions and the Role of the Diluents on the Extraction of Copper (II), Nickel (II), Cobalt (II), Magnesium (II) and Iron (II, III) with Different Types of Extractants. *Hydrometallurgy*, 78, (1-2), 92-106.
4. Cui , Y. J., Zhu Y.G., Zhai R.H., Chen D.Y., Huang Y.Z., Qiu Y.and Liang J.J.Z.(2004) Transfer of Metals from Soil to Vegetable in an Area Near a Smelter in Nanning China. *Environ. Int.*, 30, 785-791.
5. Mkadmi Y., Benabbi O., Fekhaoui M., Benakkam R., Bjjjou W., Elazzouzi M., Kadourri M., Chetouani A., (2018), Study of the Impact of Heavy Metals and Physico-chemical Parameters on the Quality of the Wells and Waters of the Holcim area (Oriental region of Morocco), *J. Mater. Environ. Sci.*, Volume 9(2), 672-679.
6. Imane Chemaou El Fehri, Abdellah El Abidi, Mohammed Fekhaoui, Abdelkbir Bellaouchou, Rachid El Jaoudi , Sabounji Abdelkader., (2016), Physical and Chemical Parameters of Groundwater area Agricultural Sidi Abdelrazak (province of khemisset), *J. Mater. Environ. Sc.*, 7 (2) 579-589.
7. Ezzaouaq, M. (2005), Hydrodynamic, Physico-chemical and Bacteriological Characterization of the Surface Waters of the Bouregrege stuary (Morocco) Subjected to Discharges from the Cities of Rabat-Salé. Thesis D.E.S. *Fac. Sci. Rabat*, 1-140.
8. Mehanned S., Chahlaoui A., Zaid A., Samih M., Chahboune M., (2014), Typologie De La Qualité Physico-chimique Del'eau Du Barrage Sidi ChahedMaroc (Typology of the physicochemical quality of the waters of the dam Sidi Chahed-Morocco) *J. Mater. Environ. Sci.*, 5(5), 1633-1642.

9. Haddad H., Ghoualem H., (2014), Characterisation Physico-chimique Des Eaux Du Bassin Hydrographique Cotier Algerois., *Larhyss Journal*, 155-167.
10. Asefers Y., Taouil H., Doubi M., Amine A., Ahmed S. Ibn., (2017) Evaluation of the Physico-Chemical Quality of the Waters of the Oulja Well in the Oum Er-Rbia Watershed., *Chemistry Research Journal*. 2(3), 144-15.
11. Asfers Y., Taouil H., Doubi M., Amine A., Ibn S., (2017) Assessment of the Physicochemical Quality of Water in the River Oum Er-Rbia- Morocco., *Journal of Applied Chemistry*, 10(5), 41-49.
12. Imane Kherrati, Alemad A., Sibbari M., Ettayea H., Ezziani K., Saidi Y., Benchikh M., Alzwi S., Chiguer H., Zgourdah Z., Bourass A., Daifi H., Elrhout O., Elkharrim K., Belghyti D., (2015),. Health Risk of Maâmora's Groundwater Pollution in Morocco, *Natural Resources*, 6, 290-305.
13. Kaveh A.R., Shahedi K., Habibnejad Roshan M., Ghorbani (2015), Assessment of Spatio-temporal Variations of Surface Water Quality and Prioritization of Pollution Sources (case Study: Talar Watershed, Mazandaran province), *Environmental Resources Research*, 3(1), 27-45.
14. Kayastha Sadhana Pradhanang., (2014),. Heavy Metal Pollution of Agricultural Soils and Vegetables of Bhaktapur District, Nepal. *Scientific Wqrd.*, 12(12), 48-52.
15. Das S. K., (2006), Small-Scale Rural Aquaculture in Assam, India – A Case Study, *NAGA, WorldFish Center Quarterly*, 29(1-2), 42-47.
16. Cui , Y.J., Zhu Y.G., Zhai R.H., Chen D.Y., Huang Y.Z., Qiu Y. and Liang J.J.Z., (2004) Transfer of Metals from Soil to Vegetable in an Area Near a Smelter in Nanning China. *Environ. Int.* 30, 785-791.
17. Sharma, R.K., M. Agrawal & F.M. Marshall. (2006); Heavy Metals Contamination in Vegetables Grown in Waste Water Irrigated Areas of Varanasi, India. *Bulletin of Environmental Contamination and Toxicology* 77: 311-318.
18. Sharma, R. K., Agrawal M. & Marshall F. M. (2007), Heavy metals contamination of soil and vegetables in suburban areas of Varanasi, India. *Ecotoxicology and Environmental Safety*, 66, 258-266.
19. Awashthi , S. K. (2000),. Prevention of Food Adulteration Act no. 37 of 1954. *Central and State Rules as Amended for 1999, Ashoka Law House, New Delhi.*, 1-71.
20. European Union. (2000). Heavy Metals in Wastes, European Commission on Environment.(http://ec.europa.eu/environment/waste/studies/pdf/heavy_metals_report.pdf).